



# Impact of Drying Methods on Physicochemical Properties, Bioactive content, and Antioxidant Activity of *Opuntia ficus-indica* Fruits

S. Belkhir <sup>\*</sup> , D. Abdessemed, I. Refas

Laboratory for the improvement of agricultural production and protection of ecosystems in arid zones (LAPAPEZA), Institute of Veterinary and Agronomic Sciences, Batna 1 University, Algeria.

## HIGHLIGHTS

- Samples dried at 60 °C retained the highest antioxidant and bioactive content.
- Betalains are present in high concentrations in both fresh pulp and dried peels.
- Dried Peels exhibit favorable characteristics with strong bioactive qualities process.

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## Abbreviations

BI=Browning Index  
CD=Convective Drying  
Deff=Effective Moisture  
Diffusivity  
DPPH=2,2-Diphenyl-1-  
Picrylhydrazyl  
DR=Drying Rate  
IC50=Half Maximal Inhibitory  
Concentration  
MR=Moisture Ratio  
SD=Shade Drying  
TA=Titrateable Acidity  
TB=Total Betalains  
TFC=Total Flavonoids  
Compounds  
TPC=Total Phenolic Compounds  
TSS=Total Soluble Solids

## ABSTRACT

**Background:** *Opuntia ficus-indica* (prickly pear), is a highly nutritious fruit known for its antioxidant and medicinal properties. However, its seasonal availability and high moisture content make it perishable, necessitating preservation methods like drying. This study aimed to assess the impact of conventional drying methods on the drying kinetics and quality characteristics of both pulp and peels of prickly pear fruit.

**Methods:** Fruits harvested in August and September 2022 were subjected two drying methods: Shade drying and Convective drying (CD) at 40 and 60 °C using a hot air oven. Physicochemical and bioactive properties were analyzed., including Total Phenolic Content (TPC) quantified via the Folin-Ciocalteu assay, and antioxidant activity measured using radical scavenging assay. Data were analyzed using XLSTAT (version 14.5.03). One-way Analysis of Variance (ANOVA) followed by Duncan test ( $p<0.05$ ), and Pearson's correlation and Principal Component Analysis for data visualization.

**Results:** Higher drying temperatures resulted in shorter drying times and lower moisture content. Drying also significantly increased the pH, while inducing changes in Total Soluble Solids and Titrateable Acidity in both pulp and peels, along with a notable color change. Peels subjected to CD at 60 °C showed the highest Total Phenolic Content and Total Betalains content, at 35.54 mg Gallic Acid Equivalents (GAE)/g dry weight and 0.991 mg/g dry weight, respectively. Flavonoid and tannin content were highest in shade-dried pulp and at 40 °C, indicating the heat sensitivity of these compounds. Minor effects on antioxidant activity ( $IC_{50}$ ) were observed in pulp during drying, compared to the peels, where the lowest  $IC_{50}$  with values around 2.10 mg/ml, was recorded for convective dried peels at 60°C.

**Conclusion:** Convective drying at 60 °C proved to be the most effective methods for drying and preserving the bioactive properties of prickly pear fruit, offering a balance between drying efficiency and retention of key nutrients and antioxidants.

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\* Corresponding author (S. Belkhir)

✉ E-mail: safia.belkhir@univ-batna.dz

ORCID ID: <https://orcid.org/0000-0002-2692-0919>

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## Introduction

In recent years, there has been a significant shift toward adopting integrative medicine and healthier dietary habits to minimize the risk of chronic diseases including diabetes, digestive disorders, cardiovascular illnesses, and cancer. This trend is driven by the high nutritional and bioactive value of natural products such as medicinal plants, fruits, and vegetables (Gallegos-Infante et al., 2009).

*Opuntia ficus-indica* is a cactus plant that thrives in arid and semiarid regions across the new and old worlds (El-Hawary et al., 2020). This species gradually gained economic importance due to its durability, resistance to harsh climatic conditions, and richness in bioactive components, as well as its nutritional value. This has led to its long-standing recognition as a functional food and medicinal plant (Barba et al., 2017). Moreover, *Opuntia* provides delicious and nutritious fruits that can be consumed fresh or processed. However, fruit processing generates substantial waste, such as prickly pear peels, which are a rich source of bioactive substances (Barba et al., 2017). Several research highlight that polyphenols and betalains extracted from prickly pear pulp and peels exhibit antioxidant, antibacterial, anti-inflammatory, and antidiabetic properties (Aruwa et al., 2018).

Like many other plants, *Opuntia* produces seasonal, perishable fruits that are susceptible to chemical and microbial deterioration due to their high moisture content, limiting their shelf life. However, employing various preservation methods such as freezing and drying can extend their shelf-life (Abrol et al., 2014).

Drying has been used since ancient times as a post-harvest preservation method for food and medicinal plants (Guine et al., 2018) to enhance their shelf life by removing most of the water content. This process significantly decreases the water activity, which in turn reduces the proliferation of germs and inhibits certain enzymatic activities. Furthermore, the reduction in moisture content decreases the overall mass of the product, making storage, packaging, and transportation more manageable and less expensive (Phuon et al., 2022). However, dehydration can negatively impact product quality by causing physicochemical damage, including Maillard reaction, color changes, oxidation, and shrinkage. Additionally, it affects the nutritional and functional components, such as phenolic compounds, vitamins, and pigments (Meng et al., 2018). Consequently, this may render the product unacceptable to consumers. The appearance and discoloration of dried products are critical physical attributes that affect consumer perceptions of food quality. Moreover, these changes can impact the availability of active compounds, such as flavonoids, carotenoids, and betalains, which are essential in natural food coloring and phytopharmaceutical industries (El-Hawary et al., 2020).

These characteristics are often influenced by various dehydration factors including temperature, airflow, drying time, and the specific drying method employed (Gallegos-Infante et al., 2009).

Various drying technologies have been used based on the type and characteristics of the product. Selecting the most appropriate drying technology and optimizing the drying conditions are critical for preservation, enhancing the added value of food, and reducing operating costs (Mohammed et al., 2020). Moreover, traditional drying methods like sun and Shade Drying (SD) are widely used for ecological and socioeconomic considerations, including ease of use, basic equipment, and low capital costs. However, this process has some inherent problems, as it is highly weather-dependent, involves long dry periods, and is susceptible to dust contamination and insect infestation. Therefore, Convective Drying (CD), also known as hot air-oven drying, is considered the most common and efficient method on the industrial scale (Shi et al., 2019). This method is more cost-effective, reduces processing time, and provides uniformity and hygiene (Abrol et al., 2014; Refas et al., 2025). However, some disadvantages are associated with this process, such as prolonged exposure to high or low temperatures during drying. Consequently, product quality declines due to the degradation of nutritional elements, color, and essential flavor (Meng et al., 2018). Existing research on prickly pear drying methods is quite limited. This study aimed to assess the physicochemical and phytochemical characteristics, as well as the antioxidant activity of both the pulp and peels of *O. ficus-indica* fruits. Additionally, it examined the impact of conventional drying methods to identify the optimal conditions for preserving the key quality attributes of the dried pulp and peels.

## Materials and methods

### *Solvents and chemicals*

2,2-diphenyl-1-picrylhydrazyl (DPPH; SPID4313-1G, Prochima-Sigma, Asie), (+)-Catechin hydrate (C1251-5G, Sigma-Aldrich, China), aluminum chloride (AlCl<sub>3</sub>; 206911-1KG, Sigma-Aldrich, Germany), ethanol (BIOCHEM Chemopharma, France), Folin-Ciocalteu's phenol reagent (BIOCHEM Chemopharma, France), gallic acid (G7384-100G, Merck, China), hydrochloric acid (HCl; 07102-2,5L-GL, Honeywell, Austria), methanol (34860-2,5L, Honeywell, France), quercetin (Q4951-10G, Merck, China), sodium carbonate (13418-1KG-R, Sigma-Aldrich, Germany), sodium hydroxide (NaOH; 30620-1KG, Honeywell, Sweden), sodium nitrite (NaNO<sub>2</sub>; 563218-25G, Merck, Germany), vanillin (V1104-100G, Sigma-Aldrich, Germany). All chemicals used were of

analytical grade.

#### Plant preparation and drying

*O. ficus-indica* fruits of the prickly pear were harvested from the Oulad Fadel region in Batna, Algeria, during August and September 2022. Fruits with consistent maturation, uniform size, and no defects were selected. First, the *Opuntia* fruits were cleaned with flowing tap water to remove dust and impurities, rinsed with distilled water, and then disinfected with ethanol (80%). Lastly, the residual moisture was removed by drying them with tissue paper. After manual peeling, the fruit pulp was cut into small cylindrical pieces each measuring 3.5 cm in diameter and 0.5 cm in thickness, and weighing  $6\pm 0.5$  g. The peels were also cut into pieces of  $2\times 4$  cm and  $3\pm 0.5$  g.

Two types of drying methods were examined to evaluate their impact on the product:

-SD: using a drying rack, the samples were placed in a shaded area at an ambient temperature (25-30 °C). The weight of the samples was monitored daily and hourly (from 09.00 to 17.00 h).

-CD: conducted using a hot-air oven (Memmert, Germany) at temperatures of 40 and 60 °C, with a constant drying air flow (1.5 m/s). The weight of the samples was monitored hourly. An analytical balance (KERN, Germany) was used to weigh samples until equilibrium was reached and a consistent weight was noted. Dehydrated samples were ground into powder, placed in an opaque airtight glass jar and kept at 4 °C until further analysis.

#### Drying curves

##### -Moisture Ratio (MR)

The MR of prickly pear is calculated according to the following equation

$$MR = \frac{X_t - X_e}{X_0 - X_e} \quad (1)$$

Where  $X_t$ : Moisture content at any time (kg water/kg dry solid),  $X_0$ : initial moisture content (kg water/kg dry solid), and  $X_e$ : the moisture content at equilibrium (kg water/kg dry solid) (Refas et al., 2025)

##### -Drying Rate (DR)

The DR of prickly pear was calculated as:

$$DR = \frac{X_{t1} - X_{t2}}{t_2 - t_1} \quad (2)$$

$X_{t1}$  and  $X_{t2}$  refer to moisture loss of the sample at two successive measurements periods  $t_1$  and  $t_2$  (Refas et al., 2025).

##### -Effective Moisture Diffusivity ( $D_{eff}$ )

$D_{eff}$  is a value that describes the rate of moisture removal. It could be expressed according to Fick's second law of diffusion for unsteady state as shown in the following formula

$$\frac{\partial X}{\partial t} = D_{eff} \nabla^2 X \quad (3)$$

Where X is the moisture content (kg water/kg db.), t is the drying time, and  $D_{eff}$  is the effective diffusivity ( $m^2/s$ ). Assuming the initial moisture content of the product at the beginning is unchanged, thermal equilibrium exists between the sample surface and the drying air, with only minimal shrinkage during drying.

The general solution of equation (3) can be obtained for common shapes such as slab for the peel (4) (Pinheiro and Castro, 2023), and a cylinder for the fruit pulp (5), using appropriate boundary conditions (Touil et al., 2014).

$$MR = \frac{X_t - X_e}{X_0 - X_e} = \frac{8}{\pi^2} \exp\left(-\frac{\pi^2 \times D_{eff}}{4L^2} t\right) \quad (4)$$

Where L is the thickness of the peel.

$$MR = \frac{X_t - X_e}{X_0 - X_e} = \frac{8}{\pi^2} \exp\left(-\frac{\varepsilon_1^2 \times D_{eff}}{r_c^2} t\right) \quad (5)$$

Where  $\varepsilon_n$  is roots of Bessel function.

#### Physicochemical proprieties

Fresh and dried fruit pulp and peels were analysed for their physicochemical characteristics. Moisture content (MC) (g/g of wet basis (wb)), pH, Titratable Acidity (TA) (g of citric acid/100 g wb), and Total Soluble Solids (TSS) (Brix°) were determined according to AOAC methods 925.10, 981.12, 942.15, and 932.12, respectively (AOAC, 2000).

#### Color measurement

The chromatic properties of fresh and dried samples were quantified using a chroma meter (Color reader CR-10, Germany). Illuminant D65 served as the light source. The CIELab color values measured included Lightness ( $L^*$ ), redness/greenness ( $a^*$ ), and yellowness/blueness ( $b^*$ ). Three replicate measurements were performed. Furthermore, the following equations were used to compute color intensity (Chroma), Hue angle ( $H^\circ$ ), and the total color change ( $\Delta E$ ):

$$Chroma = \sqrt{(a^{*2} + b^{*2})} \quad (6)$$

$$Hue\ angle = \tan^{-1}\left(\frac{b^*}{a^*}\right) \quad (7)$$

$$\Delta E = \sqrt{(a^* - a_0^*)^2 + (b^* - b_0^*)^2 + (L^* - L_0^*)^2} \quad (8)$$

$\Delta E$  represents the color difference between fresh and dried samples.  $L_0^*$ ,  $b_0^*$ , and  $a_0^*$  correspond to the color parameters of the fresh samples, while  $L^*$ ,  $a^*$ , and  $b^*$  refer to those of the dried samples.

The Browning Index (BI) is cited as an essential measure in procedures related to both enzymatic or non-enzymatic browning (Pathare et al., 2013). The BI is calculated using

$L^*$ ,  $a^*$ , and  $b^*$  values when applying the following equation:

$$BI = \frac{[100 \times (x - 0.31)]}{0.17} \quad (9)$$

Where

$$x = \frac{(a^* + 1.75 \times L^*)}{5.645 \times L^* + a^* - 3.012 \times b^*} \quad (10)$$

#### Determination of phytochemical properties

##### -Preparation of extracts

Ten g of fresh and dried pulp and peels of prickly pear fruits were combined with 100 ml of hydro ethanol solution (80% v/v) and stirred at room temperature for 24 h. The extraction was repeated twice under similar conditions. Afterwards, the filtrate was collected and the solvent was evaporated at  $40 \pm 2$  °C under decreased pressure using a rotavapor (Stuart, UK). The dried extracts were kept at 4 °C for further analysis.

##### -Total Phenolic Compounds (TPC)

The Folin–Ciocalteu method, described by Singleton and Rossi (1965), was employed to assess the TPC in the extracts. Five tenths ml of crude extract (3 mg/ml) was mixed with 2.5 ml of diluted Folin–Ciocalteu reagent. After five min in the dark, 0.8 ml of a 7.5% sodium carbonate solution was added. A spectrophotometer (UV-visible 7305– Jenway, UK) was used to detect the absorbance at 765 nm following a two h dark incubation. Triplicate measurements were taken and the results were expressed as mg Equivalent of Gallic Acid (EGA)/g of dry weight using gallic acid standard curve.

##### -Total Flavonoids Compounds (TFC)

TFC was assessed according to the method outlined by Zhishen et al. (1999). In this procedure, 0.5 ml aliquot of the crude extract solution was combined with 1.5 ml of distilled water and 150 µl of  $\text{NaNO}_2$ . After 5 min, 150 µl of  $\text{AlCl}_3$  (10%) and 0.5 ml of NaOH were added. The mixture was then incubated for 30 min at room temperature in the dark. Absorbance was read at 430 nm by a spectrophotometer. Quercetin served as the standard, and TFC was expressed as mg equivalent of Quercetin (QE)/g of dry weight.

##### -Tannins compounds

The tannins content was measured using the vanillin/HCl method from Price et al. (1978). The vanillin reagent, freshly prepared before the experiment, consisted of equal parts of methanol solutions containing 8% HCl and 1% vanillin. An aliquot of three ml of the vanillin reagent was added to 0.6 ml of the sample. After a 20 min incubation at 30 °C, the mixture's absorbance was measured at 500 nm.

Catechin served as the reference standard, and the TFC was expressed as mg Equivalent of Catechin (EC)/100 g of dry weight.

##### -Betalains content

The quantification of betalains pigments, particularly betaxanthin and betacyanin, was performed using the methodology described by Nadia et al. (2013). This involved using 80% methanol as the extraction solvent. The samples were stirred for 6 h and then centrifuged at 4,000 rpm for 20 min. The betalains concentrations were determined by measuring spectrophotometer absorbance at 480 nm and 535 nm, respectively. The results were expressed according to the following equation:

$$B(\text{mg/g}) = \left( \frac{A \times DF \times M \times V}{\epsilon \times W \times L} \right) \times 100 \quad (11)$$

Where A: absorbance, V: extract volume, W: sample masse (g), L: cuvette length (one cm), DF: dilution factor, M: molecular weight molar,  $\epsilon$ : extinction coefficient, (M: 550 g/mol,  $\epsilon$ : 60,000 L/mol.cm for betacyanin), and (M: 308 g/mol,  $\epsilon$ : 48,000 L/mol.cm for betaxanthin).

##### DPPH antioxidant assay

The Free radical scavenging activity of prickly pear pulp and peels extracts was evaluated using the DPPH assay, as outlined by Brand-Williams et al. (1995). A one ml aliquot of various dilutions was mixed with two ml of DPPH (0.1 mmol). The process was performed in triplicate. The reaction mixture was vortexed for 30 s and left to stand in the dark for 20 min at room temperature. Using a spectrophotometer, the absorbance was measured at 517 nm against a blank. Total Antioxidant Capacity (TAC) was expressed as the percentage inhibition of the DPPH radical, calculated using the following equation:

$$TAC \% = \left( 1 - \frac{Abs_{\text{sample}}}{Abs_{\text{control}}} \right) \times 100 \quad (12)$$

Where;  $Abs_{\text{sample}}$  is the absorbance of the mixture of the extract and reagent and  $Abs_{\text{control}}$  is the absorbance of the blank.

The half maximal Inhibitory Concentration ( $IC_{50}$ ), representing the concentration at which 50% of the DPPH radicals are scavenged by the diluted samples, was determined.

##### Statistical analysis

The statistical analysis of all experiments was conducted using XLSTAT (version 14.5.03.). The results were provided as Mean  $\pm$  SD. An analysis of variance ANOVA was conducted with Duncan's post hoc test to assess the differences among means at a 95% confidence interval ( $p < 0.05$ ). The correlation between bioactive components

and antioxidant activity assays was analysed using Pearson's test. A biplot Principal Component Analysis (PCA) was used to visualise the physicochemical properties and bioactive components, as well as the IC<sub>50</sub> DPPH of fresh and dried prickly pear fruit pulp and peels in a 2D representation.

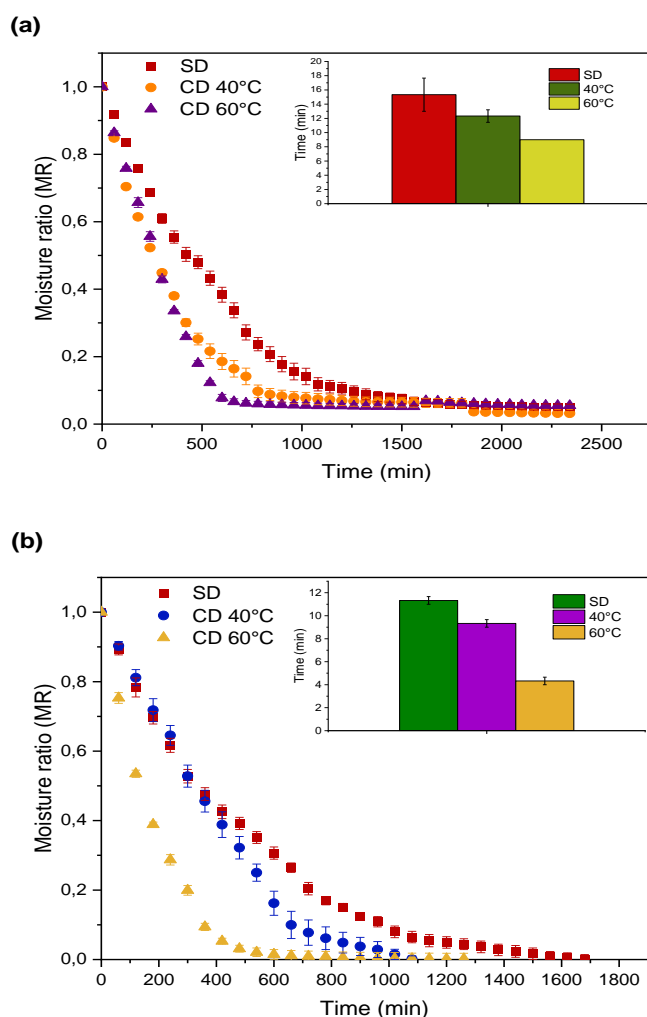
## Results and discussions

### Drying curves of prickly pear pulp and Peels fruit

-MR

Figure 1 depicts the changes in MR during CD at 40 and

60 °C and SD of prickly pear fruit pulp and peels. The MR decreased exponentially throughout drying; a behaviour consistent with the dehydration of fruits and vegetables (Onwude et al., 2016). As anticipated, higher temperatures reduced drying times, primarily due to increased vapour pressure within the fruit, accelerating moisture removal (Touil et al., 2014). These findings align with previous reports by Refas et al.(2025). Hot air drying at varying temperatures shortened the total drying duration from 15 to 9 h for prickly pear pulp, and from 11 to 4 h for peels compared to SD. Notably, peel drying progressed more rapidly than pulp drying.

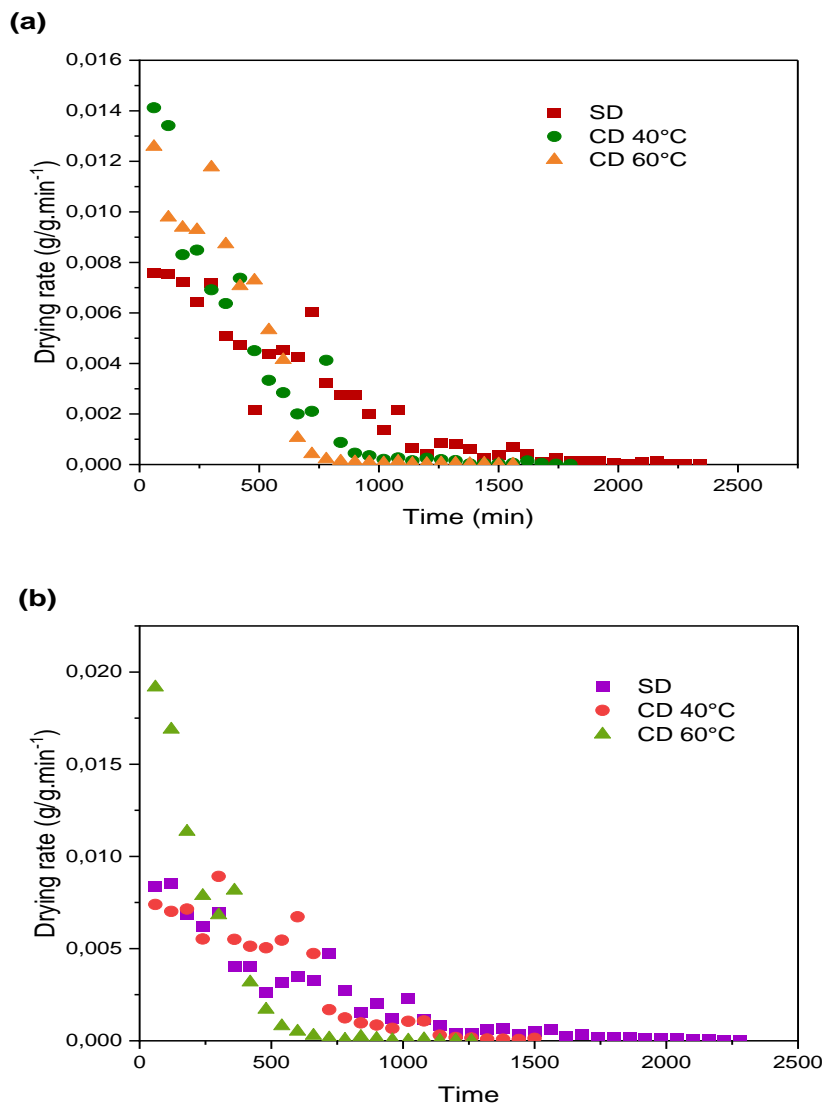


**Figure 1:** Moisture Ratio (MR) of prickly pear pulp (a) and peels (b) during Shade Drying (SD) and Convective Drying (CD)

-DR

Figure 2 illustrates the DR of prickly pear fruit pulp and peels subjected to CD and SD. Initially, the DR increased significantly, likely due to rapid heat absorption by the large amount of water within the fresh fruit and peels

(Sadin et al., 2014). However, the rates gradually declined until the drying process concluded. Elevated temperatures improved both the DR and diffusion coefficient, thereby accelerating water evaporation compared to SD (Figure 2).



**Figure 2:** Drying Rate (DR) of prickly pear pulp (a) and peels (b) during Shade Drying (SD) and Convective Drying (CD) at different temperatures

$-D_{eff}$

The  $D_{eff}$  for convective and shade dried pulp and peels of prickly pear fruit were determined using Equation (4) and (5). The results, presented in Table 1, reveal an insignificant statistical difference in  $D_{eff}$  values. Notably, lower  $D_{eff}$  values were observed for SD, while higher values were found for CD, indicating slower moisture migration in SD due to the absence of forced airflow and lower drying temperatures. These values denote the longer drying time required for this conventional methods. During

CD, the  $D_{eff}$  increased with rising temperatures, especially for the pulp, as higher thermal energy enhanced water molecule mobility and accelerated moisture removal. Notably, the peels exhibited lower diffusivity values under convective conditions, suggesting a more resistant structure that retained moisture longer than the pulp. These findings align with previous reports by Aral and Beşe (2016) and support the faster dehydration of peels observed in the MR section.

**Table 1:** Determination of Effective Moisture Diffusivity ( $D_{eff}$ ) for convective and shade dried pulp and peel of prickly pear fruit

	Pulp			Peels		
	SD	CD 40 °C	CD 60 °C	SD	CD 40 °C	CD 60 °C
$D_{eff}$ (m <sup>2</sup> /s)	1.37.10 <sup>-9</sup>	1.57.10 <sup>-9</sup>	1.96.10 <sup>-9</sup>	5.39 .10 <sup>-9</sup>	7.76.10 <sup>-11</sup>	7.08.10 <sup>-11</sup>
	±	±	±	±	±	±
	3.39.10 <sup>-10a</sup>	3.39.10 <sup>-10a</sup>	3.39.10 <sup>-10a</sup>	1.55.10 <sup>-11a</sup>	5.84.10 <sup>-12a</sup>	2.68.10 <sup>-11a</sup>

Values are the Mean±SD. Different letters present significant differences ( $p \leq 0.05$ ).  
SD=Shade Drying; CD=Convective Drying at 40 and 60 °C.

### Physicochemical proprieties

Table 2 shows the initial moisture content in *O. ficus-indica* fresh pulp and peels with values of 0.85 and 0.82 (g H<sub>2</sub>O/g dry weight), respectively. These values align with those reported by Nadia et al. (2013) for *O. ficus-indica* varieties grown in northern Algeria. Following drying, the moisture content significantly decreased ( $p < 0.05$ ). Notably, CD at 60 °C and SD were most effective in significantly reducing moisture due to high temperature and prolonged drying time, respectively. The pH values, Titrable Acidity (TA), and TSS all increased significantly during the drying process ( $p < 0.05$ ). The rise in TA can be attributed to the

reduction in moisture, causing higher concentrations with the production of acids by converting sugars and other chemical reactions (Abrol et al., 2014). Lower moisture content also increased dry matter concentration (Bourhia et al., 2020), resulting in a marked TSS increase. The CD at 60 °C samples had the highest content of soluble solids, with 58.33 and 60 °Brix for pulp and peels, respectively. These values were lower than those measured by Ettalibi et al. (2020) for Moroccan varieties, and higher than those reported by Nadia et al. (2013) and García-Cayuela et al. (2019) in yellow prickly pear pulp of Algerian and Spanish varieties, respectively.

**Table 2:** Physicochemical and color characterisation of fresh and dried Pulp and Peels of *Opuntia ficus-indica* fruit

Physico-chemical characteristics		Drying methods			
		*F	**SD	***CD 40°C	CD 60°C
Moisture content (g H <sub>2</sub> O/g dry weight)	Pulp	0.847±0.021 <sup>b</sup>	0.050±0.015 <sup>a</sup>	0.056±0.022 <sup>a</sup>	0.049±0.012 <sup>a</sup>
	peels	0.825±0.016 <sup>b</sup>	0.036±0.021 <sup>a</sup>	0.040±0.043 <sup>a</sup>	0.025±0.006 <sup>a</sup>
pH	Pulp	5.87±0.07 <sup>a</sup>	6.59±0.05 <sup>c</sup>	6.14±0.01 <sup>b</sup>	5.91±0.06 <sup>a</sup>
	peels	4.90±0.02 <sup>a</sup>	5.37±0.03 <sup>b</sup>	5.88±0.02 <sup>c</sup>	5.73±0.16 <sup>c</sup>
TA (g citric acid/100 g w <sub>b</sub> )	Pulp	0.16±0.02 <sup>a</sup>	0.22±0.02 <sup>b</sup>	0.28±0.03 <sup>c</sup>	0.25±0.03 <sup>bc</sup>
	peels	0.27±0.02 <sup>a</sup>	0.35±0.01 <sup>b</sup>	0.31±0.03 <sup>ab</sup>	0.46±0.04 <sup>c</sup>
TSSs (Brix°)	Pulp	13.50±0.5 <sup>a</sup>	55±5 <sup>c</sup>	45±5 <sup>b</sup>	58.33±7.64 <sup>c</sup>
	peels	13.17±0.5 <sup>a</sup>	56.67±7.64 <sup>b</sup>	55.00±5 <sup>b</sup>	60.00±5 <sup>b</sup>
<b>Color parameters</b>					
	Lightness (L*)	Pulp	49.13±0.42 <sup>c</sup>	36.70±1.51 <sup>a</sup>	38.23±1.58 <sup>ab</sup>
Redness (a*)	peels	40.37±1.05 <sup>a</sup>	36.63±4.57 <sup>a</sup>	37.20±2.55 <sup>a</sup>	38.03±1.66 <sup>a</sup>
	Pulp	18.97±7.05 <sup>b</sup>	13.73±1.93 <sup>ab</sup>	10.23±0.90 <sup>a</sup>	12.87±0.15 <sup>ab</sup>
Yellowness (b*)	peels	10.43±0.38 <sup>a</sup>	17.57±1.80 <sup>bc</sup>	15.43±1.75 <sup>b</sup>	19.66±2.20 <sup>c</sup>
	Pulp	43.13±7.48 <sup>a</sup>	37.87±0.15 <sup>a</sup>	39.57±3.65 <sup>a</sup>	35.63±0.60 <sup>a</sup>
Chroma	peels	19.30±0.25 <sup>a</sup>	27.83±4.25 <sup>b</sup>	29.63±2.51 <sup>b</sup>	31.80±1.32 <sup>b</sup>
	Pulp	47.39±8.24 <sup>b</sup>	40.31±0.61 <sup>ab</sup>	40.90±3.34 <sup>ab</sup>	37.89±0.55 <sup>a</sup>
Hue angle (H°)	peels	21.80±0.11 <sup>a</sup>	32.98±3.87 <sup>b</sup>	33.45±2.22 <sup>b</sup>	37.43±1.49 <sup>b</sup>
	Pulp	66.31±7.21 <sup>a</sup>	70.10±2.63 <sup>ab</sup>	75.37±2.40 <sup>b</sup>	70.14±0.45 <sup>ab</sup>
Color change (ΔE)	peels	61.39±1.15 <sup>a</sup>	57.53±4.29 <sup>a</sup>	62.44±3.62 <sup>a</sup>	58.30±3.22 <sup>a</sup>
	Pulp	-	16.46±3.37 <sup>a</sup>	15.98±6.28 <sup>a</sup>	14.97±5.05 <sup>a</sup>
Browning index (BI)	peels	-	12.87±2.12 <sup>a</sup>	12.22±2.38 <sup>a</sup>	16.11±1.22 <sup>b</sup>
	Pulp	192.34±49.92 <sup>b</sup>	247.94±23.60 <sup>ab</sup>	242.67±23.20 <sup>ab</sup>	186.06±8.58 <sup>a</sup>
	peels	81.96±2.60 <sup>a</sup>	160.86±19.22 <sup>b</sup>	168.95±34.24 <sup>b</sup>	184.52±15.50 <sup>b</sup>

Values are the Mean±SD. Different letters in row present significant differences ( $p \leq 0.05$ ) for each parameter. TA=Titrateable Acidity; TSS=Total Soluble Solid; \*F=Frech; \*\*SD=Shade Drying; \*\*\*CD=Convective Drying at 40 and 60°C.

### Color characterisation

Color is a key quality indicator that affects the commercial value and consumer acceptance of food products, serving as a sign of ripeness and the impact of treatment. The CIELab colour parameters are commonly used to track color changes during the drying process of products. Table 2 shows a significant decrease ( $p < 0.05$ ) in Lightness ( $L^*$ ) for fresh pulp and peel, indicating a darker color. The  $L^*$  value decreased by 25.30, 22.18, and 17.84% for pulp and 28.17, 19.28, and 12.38% for peel after SD, and CD at 40 and 60 °C, respectively; suggesting that higher temperatures increase lightness, making it closer to fresh samples, possibly due to the Maillard reaction, enzymatic browning, or pigments degradation (Guclu et al., 2022). Consequently, shorter drying times reduce the duration of oxidation of some phenolic components and the intensity of browning reactions (Guine et al., 2018).

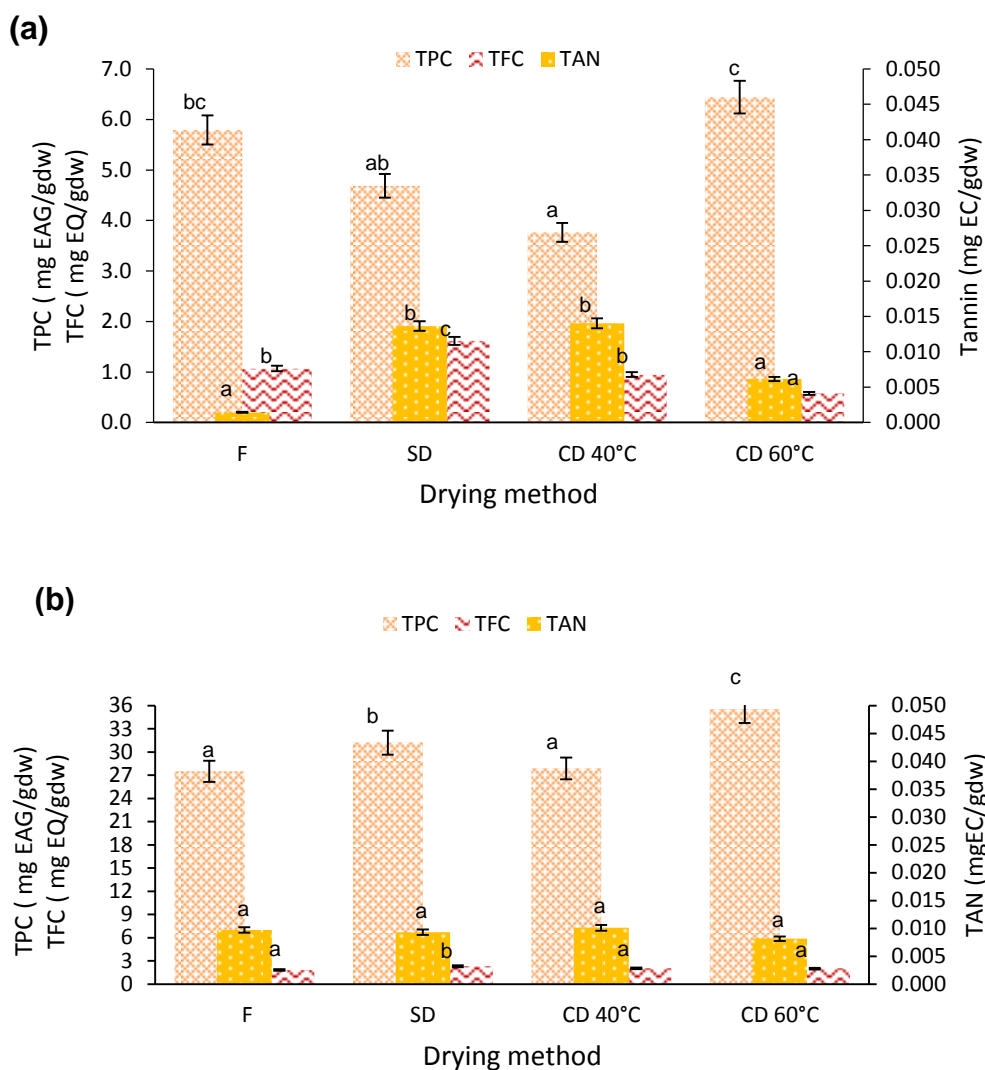
The redness value ( $a^*$ ) revealed a significant reduction in dried pulp, with the CD at 40 °C sample being the lowest, showing a decrease of 46.06%. Conversely, the redness increased significantly ( $p < 0.05$ ) in dried peels, reaching a peak of 46.97% in the 60 °C sample. Changes in  $a^*$  value reflect the degradation or concentration of red pigments, consistent with betacyanin levels in dried *Opuntia* fruit (Figure 4). The yellow pigment ( $b^*$ ) decreased significantly in pulp (17.38% at 60 °C), while it increased ( $p < 0.05$ ) by 39.84% in peels. These results align with the betaxanthin yellow pigment levels shown in Figure 4. The Chroma and Hue° values for fresh pulp were 47.39 and 66.31, respectively. Variations in drying temperature led to significant decreases in Chroma and increases in Hue angle, findings consistent with those reported by Bourhia et al. (2020), and Phuon et al. (2022). For the *Opuntia* fruit peels, there was no statistical difference in Hue° values between fresh and dried samples, but Chroma values increased significantly ( $p < 0.05$ ), indicating greater color intensity compared to fresh peels. The color change  $\Delta E$  was most prominent in pulp dried with SD and peels dried with CD at 60 °C, illustrating the browning process. The BI was calculated to confirm these findings, with the highest values observed under the same conditions. This may be attributed to the high amount of reducing sugars (Hernández García et al., 2020), as well as the oxidation of ascorbic acid during heat drying (Ettalibi et al., 2020), which enhance the Maillard reaction. The BI values for fresh pulp and peels were 192.34 and 81.96, respectively; linked to high moisture content promoting enzymatic browning through oxidation of phenolic compounds to o-quinones by Polyphenolic Oxidase (PPO), which then polymerizes to generate dark-colored pigments (Yap et al., 2020). Pathare et al. (2013) noted that color changes during drying are often non-uniform due to varying

moisture loss rates and chemical reactions in different food parts.

### Phytochemical proprieties

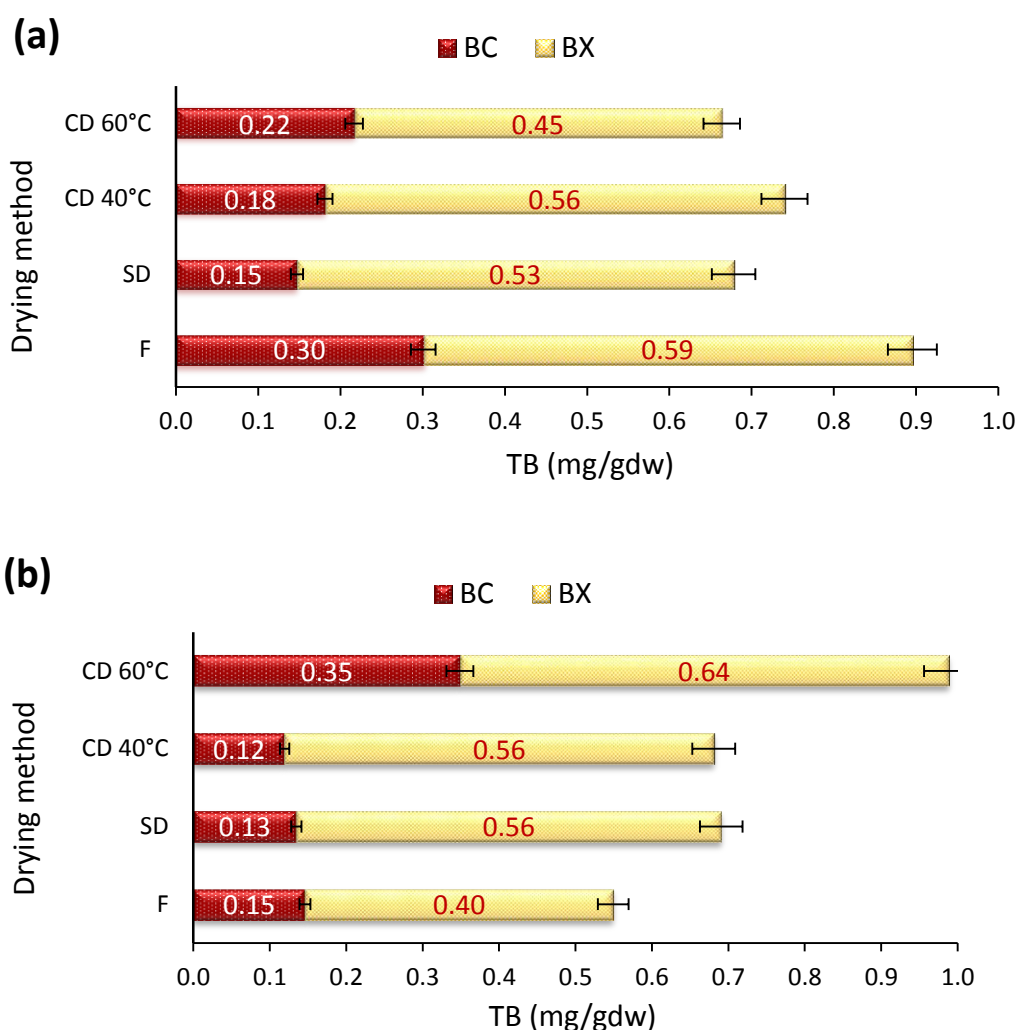
Figure 3 shows the impact of drying methods on TPCs, TFCs, and tannin in the prickly pear pulp and peels. The TPC in fresh pulp and peels was 5.79 and 27.49 mg Gallic Acid Equivalents (GAE)/g dry weight, respectively; A significant decrease ( $p < 0.05$ ) in TPC was observed in the pulp after SD and convective drying at 40 °C. In contrast, TPC increased in both the pulp and peels convective dried at 60 °C, as well as in the peels after SD and CD at 40 °C. As reported by Farahmandfar et al. (2020) for dried orange peels and Kaur et al. (2020) for dried tomato and sweet pepper, the decline of the TPC can be attributed to enzymatic degradation and polyphenol-protein interactions. Furthermore, the prolonged drying duration in SD and CD 40 °C creates favorable conditions for enzymatic oxidation of polyphenol. Additionally, the TPC reduction may also be attributed to polyphenol binding to other compounds; such as proteins, or to changes in its chemical structure, making them unextractable or undetectable by the analytical methods employed (Vega-Gálvez et al., 2009). These results are similar to those reported by El Mannoubi (2021) and Ettalibi et al. (2020). The CD at 60 °C led to a higher content of total polyphenol in both pulp and peels of the *Opuntia* fruit at 6.44 and 35.54 mg GAE/g dry weight. These findings are consistent with those reported by Capar et al. (2023). As clarified by Capar et al. (2023) and Guclu et al. (2022), the increase in TPC throughout the drying process is linked to the generation of novel liberated phenolic fragments. This process involves the release of phenolic constituents from their bonded state, facilitated by heat-induced breakdown of ester bonds between phenolic and the cell wall.

The production of Maillard reaction products during heat exposure may contribute to the generation of new phenolic compounds from their precursors at elevated temperatures (Sultana et al., 2012). Low TPC in the fresh sample may be attributed to its high moisture content, which increases enzymatic reactions that cause the degradation of phenolic components (Hossain et al., 2010). As described in Figure 3, SD showed the highest TFC of 1.615 and 2.303 mg QE/g dry weight for pulp and peels, respectively, followed by CD at 40 °C then at 60 °C. These findings indicate that lower drying temperatures may have advantages in preserving heat-sensitive bioactive compounds (Farahmandfar et al., 2020). Additionally, the tannin content significantly increased in samples dried at lower temperatures compared to fresh samples, consistent with the results reported by Najman et al. (2023) for dried quince fruit.



**Figure 3:** Effect of different drying methods on the phenolic components of prickly pear pulp (a) and peels (b).

Means with the different letter present a significant difference at ( $p < 0.05$ ). TPC=Total Phenolic Components; EAG=Equivalent of Gallic acid; TFC=Total Flavonoid Component; EQ=Equivalent of Quercetin; TAN=Tannin; EC=Equivalent of Catechin; F=Fresh; SD=Shade Drying; CD=Convective Drying at 40 and 60 °C



**Figure 4:** Effect of different drying methods on the betalain content of prickly pear fruit pulp (a) and peel (b). TB=Total Betalains; BC=Betacyanins; BX=Betaxanthins; F=Fresh; SD=Shade Drying; CD=Convective Drying at 40 and 60 °C

#### Betalains content

Figure 4 illustrates the Total Betalains (TB) content which represents the sum of the betacyanins and betaxanthin. A statistically significant difference ( $p < 0.05$ ) was detected between fresh and dried samples. The fresh pulp contained the highest TB content (0.894 mg/g dry weight), which is higher than the values reported by García-Cayuela et al. (2019) and Nadia et al. (2013) but lower than findings of Cejudo-Bastante et al. (2014).

The drying process led to a reduction in the TB content of the pulp, with values of 0.747 mg/g dry weight for SD, 0.706 mg/g dry weight for CD at 40 °C, and 0.691 mg/g dry weight for CD at 60 °C. This may be attributed to thermal degradation and the loss of sensitive compounds during the drying process. Conversely, a significant increase in TB was observed in dried *O. ficus-indica* peels with the highest amount (0.991 mg/g dry weight) found in CD samples at 60 °C, followed by SD and CD at 40 °C

(0.691 and 0.681 mg/ml, respectively), compared to fresh peels (0.549 mg/ml). Similar results were reported by Ettalibi et al. (2020). The increased TB content in dried peels, especially at 60 °C, suggests that higher drying temperatures can preserve and enhance these compounds. This may be due to the short exposure time to heat and the concentration effect as moisture is removed.

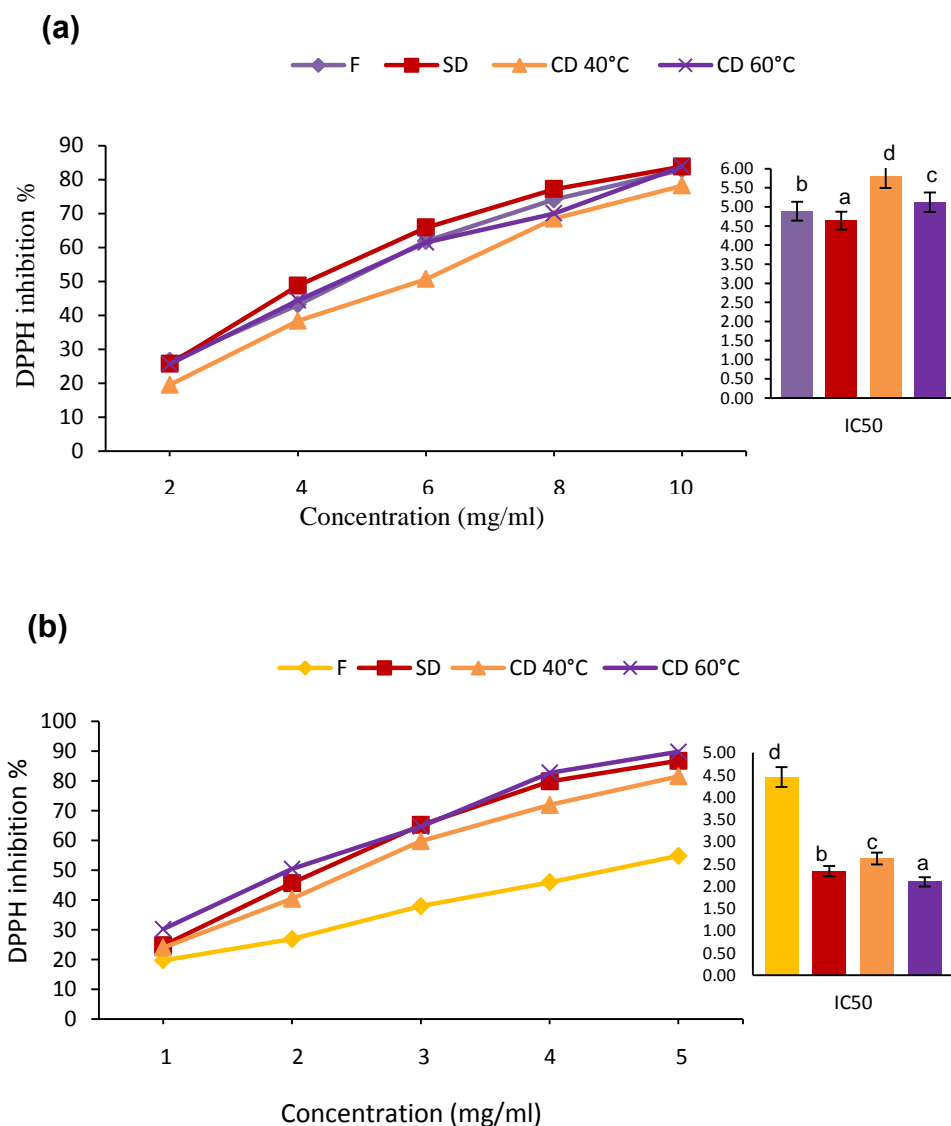
#### Antioxidant activity

The amount of sample required to achieve a 50% reduction in the initial DPPH concentration ( $IC_{50}$ ) serves as a critical criterion for evaluating antioxidant activity, with lower  $IC_{50}$  values indicating stronger antioxidant capacity. As illustrated in Figure 5, significant differences ( $p < 0.05$ ) were observed in the antioxidant capacities of fresh and dried *Opuntia* pulp and peels.

The lowest  $IC_{50}$  value for dried pulp was found in the shade-dried sample, with a value of 4.64 mg/ml, followed

by 4.88 mg/ml in the fresh sample. Conversely, dried peels exhibited a marked increase in antioxidant activity compared to fresh peels. The best antioxidant activity for the dried peels was recorded in the CD samples at 60 °C,

which had an IC<sub>50</sub> of 2.10 mg/ml. Dehydration at elevated temperatures exhibited greater antioxidant activity compared to lower temperatures. These findings align with those reported by Sultana et al. (2012).



**Figure 5:** Effect of different drying methods on the antioxidant activity of various concentrations and Half Maximal Inhibitory Concentration (IC<sub>50</sub>) of prickly pear pulp (a) and peels (b) extracts.

Different letters in IC<sub>50</sub> graph bars present a significant difference at ( $p < 0.05$ ). F=Fresh; SD=Shade Drying; CD=Convective Drying at 40 and 60 °C

Nicoli et al. (1999) stated that the retention or loss of bioactive molecules after treatment, as a result of chemical, enzymatic, or thermal impact, can lead to either an increase or decrease in antioxidant activity. The findings suggest that prickly pear peels contains higher amounts of phenolic compounds and greater antioxidant activity than the pulp, aligning with previous reports by El Mannoubi (2021). Alkaltham et al. (2020) reported that plants commonly create

secondary metabolites as response to stress. The synthesis of these substances in plants is known to occur in response to stress factors such as physical injury, pathogen invasion, extreme temperatures, and dehydration. Therefore, as a protective barrier, the peels accumulates higher levels of phenolic compounds than the pulp, enhancing its resistance to environmental stress.

**Table 3:** Correlation of Bioactive content with Half Maximal Inhibitory Concentration (IC<sub>50</sub>) 2,2-Diphenyl-1-Picrylhydrazyl (DPPH) Inhibition of pulp and peels extracts

	Pearson's correlation Coefficient (r)					
	TPC	TFC	Tannins	Betacyanin	Betaxanthin	TB
IC <sub>50</sub> pulp	-0.423	-0.543	0.296	-0.6123*	0.0759	-0.3294
IC <sub>50</sub> peel	-0.691*	-0.576	0.193	-0.405	-0.898**	-0.730**

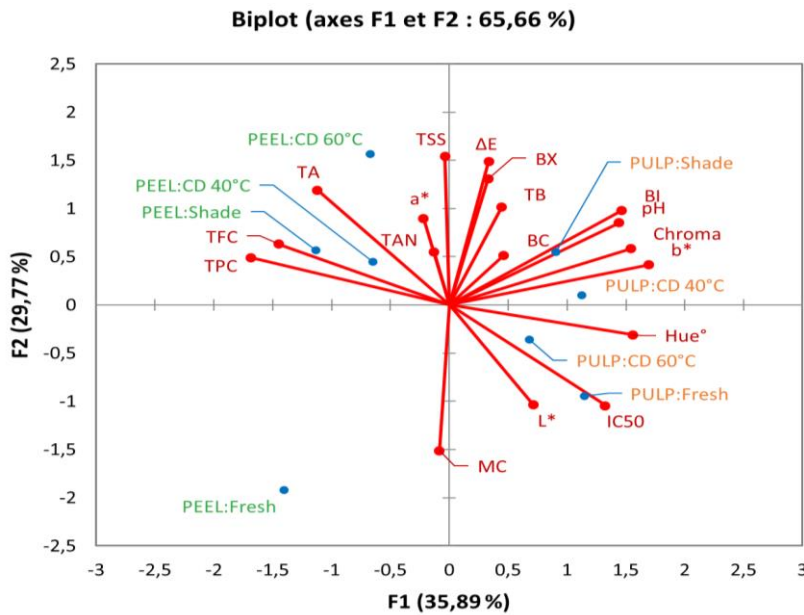
\*Correlation is significant at  $p < 0.05$

\*\*Correlation is significant at  $p < 0.01$

TB=Total Betalains; TFC=Total Flavonoids Compounds; TPC=Total Phenolic Compounds

Table 3 demonstrates the correlation between bioactive components and the antioxidant capacity IC<sub>50</sub> of *O. ficus-indica* pulp and peels. A negative correlation was observed between TPC, TFC, and betacyanin, with IC<sub>50</sub> in pulp extracts with values of -0.423, -0.543, and -0.6123, respectively. This suggests that, in addition to TFC, betacyanin plays a significant role in influencing the antioxidant activity of various pulp extracts. For the peels extracts, a negative correlation is observed between TPC, TFC, betacyanin, and betaxanthin, with the strongest

negative correlation found between betaxanthin and IC<sub>50</sub> (coefficient of -0.898), followed by TPC (-0.691), TFC (-0.576), BC (-0.405). The lower correlation between tannin compounds and IC<sub>50</sub> in both pulp, and peels extracts may be due to the the relatively low concentration of tannins in these extracts. Overall, correlation analysis indicated that TPC, TFC, and TB significantly influence the DPPH IC<sub>50</sub> inhibition values of fruit pulp and peels extracts. This finding is consistent with the results reported by El Mannoubi (2021) and Sultana et al. (2012).



**Figure 6:** Principal Component Analysis (PCA) of different dried product of prickly pear pulp and peel on the physico-chemical proprieties, bioactive, and antioxidant content  
 E: Color change; a\*: Redeness; b\*: Yelessness; BC: Betacyanin; BI: Browning Index; BX: Betaxanthin; CD: Convective Drying; DR: Drying Rate; L\*: Lightness; MC: Moisture Contenant; TA: Titratable Acidity; TAN: Tannins; TB: Total Betalains; TFC: Total Flavonoids Compounds; TPC: Total Phenolic Compounds; TSS: Total Soluble Solids

PCA illustrated in Figure 6, shows that the first two principal components, F1 and F2, account for 65.66% of the total variance. The distribution of samples along these components highlights significant compositional differences between fresh and dried samples, as well as between peels and pulp samples. The data indicate that Chroma, Hue°, IC<sub>50</sub>, and other variables on the right side are strongly associated with the F1 component. Whereas, TFC, TPC, and TA are negatively associated with F1 component. The close clustering of dried peels, particularly those dried by SD and CD at 40 °C, suggests that these samples share similar physico-chemical, phytochemical, and antioxidant properties. Furthermore, the dried peels samples show a strong positive correlation with TPC, TFC, and tannins, indicating higher retention of bioactive compounds; while their negative correlation with IC<sub>50</sub> suggests enhanced antioxidant activity. The clear separation between fresh and dried peels underscores the significant impact of drying on peels composition. In contrast, pulp samples appear more dispersed, suggesting that different drying methods can alter their properties. These samples are primarily associated with IC<sub>50</sub>, pH, and color parameters (L\*, Chroma, Hue°, and BI), as well as TB, reflecting their sensitivity to heat-induced changes in pigmentation and antioxidant capacity.

### Conclusion

This study examined the effect of conventional drying methods on the physico-chemical and phytochemical properties, as well as the antioxidant activity of *O. ficus-indica* pulp and peels. The results indicate that CD at 60 °C and SD are the most effective methods for preserving these properties. Notably, the peels contains a higher concentration of phenolic compounds and exhibits greater antioxidant activity compared to the pulp. Moreover, the peels undergoes favourable transformations during drying, effectively retaining its beneficial properties. These findings highlight the potential for optimizing conventional drying techniques and leveraging prickly pear fruits, particularly peels, in the food and pharmaceutical industries.

While the drying methods employed in this study are effective and economically viable, further research is needed to optimize their application. Specifically, a more extensive exploration of drying techniques and conditions, including temperature and airflow variations, is required to gain a deeper understanding of their impact on the properties of *O. ficus-indica* fruit.

### Author contributions

S.B. designed the work, collected the samples, performed the experimental work, analysed and interpreted the data,

and wrote the manuscript; D.A. designed and supervised the work; I.R. analysed and interpreted the drying kinetics data. All authors read and approved the final manuscript.

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### Conflicts of interest

The authors declare no conflict of interest.

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### Ethical consideration

Not applicable.

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